

IDENTIFYING SOURCES OF STRAY CARBON DIOXIDE GAS CONTAMINATION IN WESTERN PENNSYLVANIA

**Christopher D. Laughrey, *Pennsylvania
Geological Survey***

**Fred J. Baldassare, *Pennsylvania
Department of Environmental Protection***

CO₂ contamination of building spaces is an emerging environmental problem in western Pennsylvania.



CO₂ contamination of building spaces

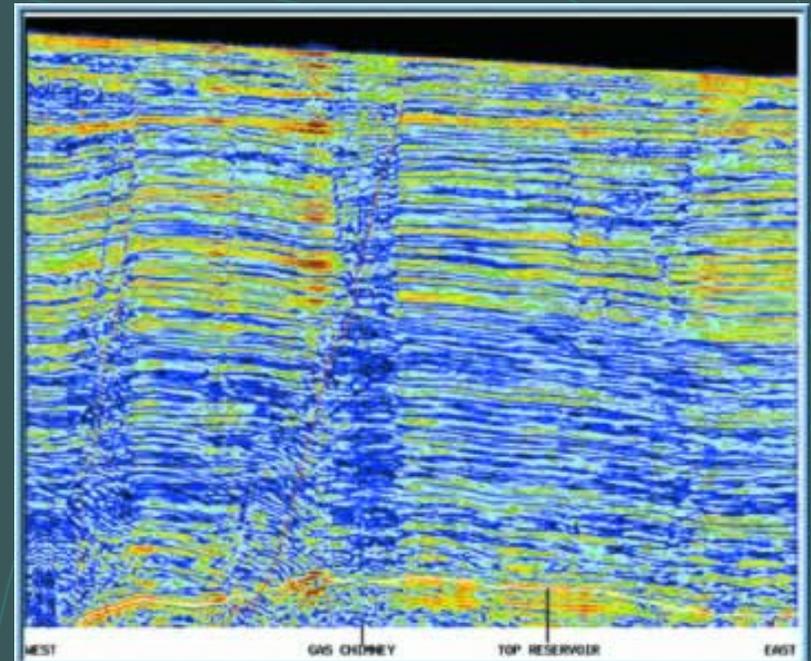
- Ambient atmosphere = 0.03%
 - TLV = 0.5%
 - TLV-STEL=1.5%
 - Humans lose consciousness in atmosphere containing 10% or more CO₂; respiratory paralysis and death may follow
 - At lower concentrations: Increased pulse and breathing rates, clumsiness, headaches, and dizziness.
- In some instances:
 - CO₂ concentrations in excess of 25%
 - Corresponding O₂ concentrations less than 10%
 - Dangerous and potentially lethal



Potential sources of stray CO₂ in western Pennsylvania

- Geologic sources
- Anthropogenic sources

Potential geologic sources *(unlikely)*



Potential anthropogenic sources

- Fossil fuel combustion
- Lime and cement manufacture
- Beverage Carbonation
- Waste combustion
- Soda ash manufacture
- CO₂ consumption and use
- Landfills
- Limestone and dolostone use
- Coal mining activities



CO₂ from abandoned deep mines in western Pennsylvania

- “Blackdamp”
- Usually associated with communication of abandoned mine atmosphere with surface structures via old shafts or fractured bedrock
- 12 – 15 % CO₂ in abandoned mine atmosphere
- Driven in large part by changes in barometric pressure

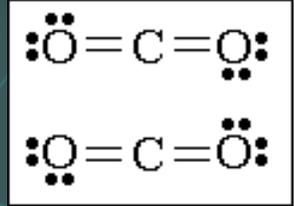


CO₂ from surface mines

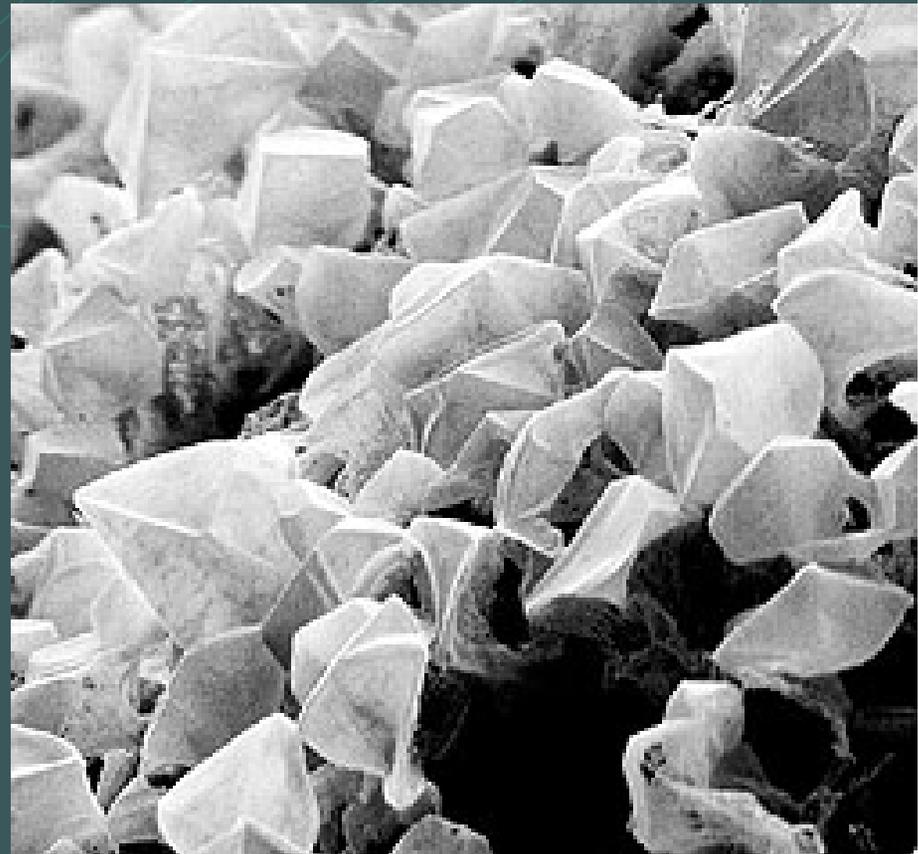
- Abandoned mines
- Reclaimed mines
- Active mines

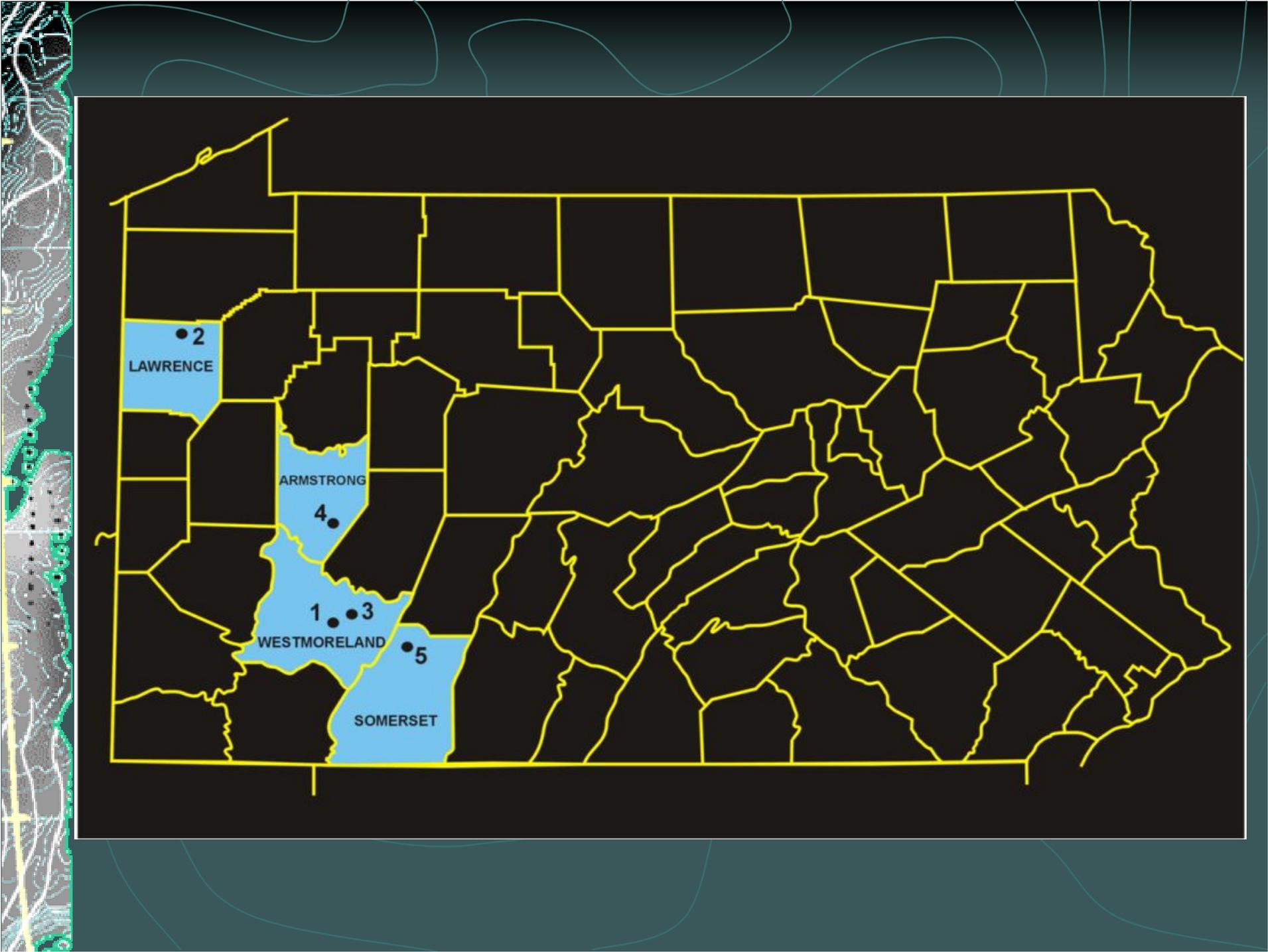


CO₂ contamination in western Pennsylvania



- Sources are not evident
- Utility of isotope geochemistry
- Impact on individuals and communities
- Potential problems with subsurface carbon sequestration





CO2 in basements, laundry rooms, game rooms



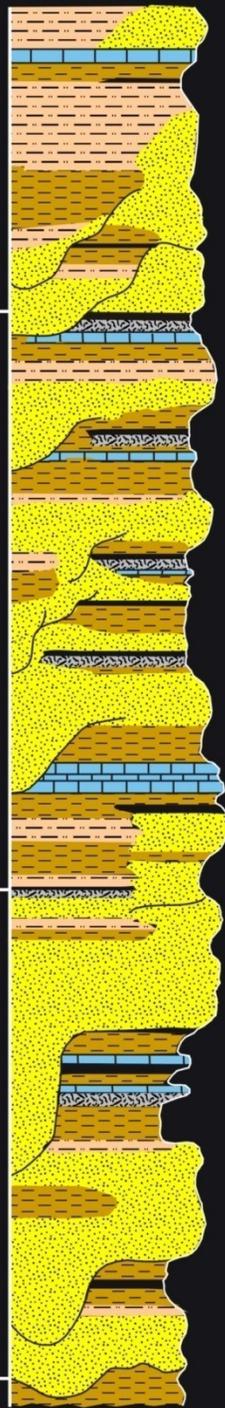


PENNSYLVANIAN

POTTSVILLE GR.

ALLEGHENY GR.

CONEMAUGH GR.



Brush Creek limestone and coal

Mahoning coal

Upper Freeport coal

Lower Freeport coal

Upper Kittanning coal

Upper Worthington sandstone

Middle Kittanning coal

Lower Kittanning coal

Vanport Limestone

Clarion coal

Brookville coal

Upper Mercer coal

Lower Mercer coal

Quakertown coal

 Sandstone

 Siltstone

 Shale

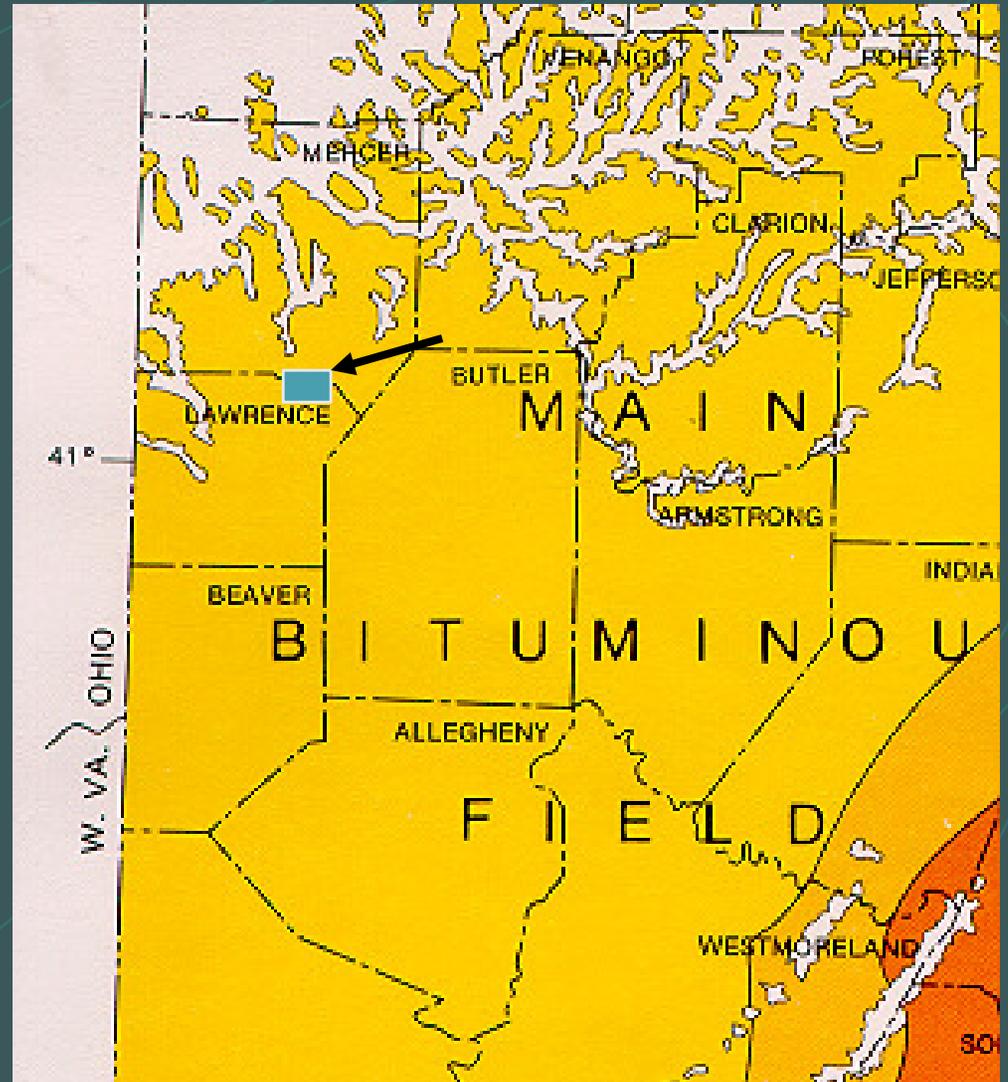
 Claystone

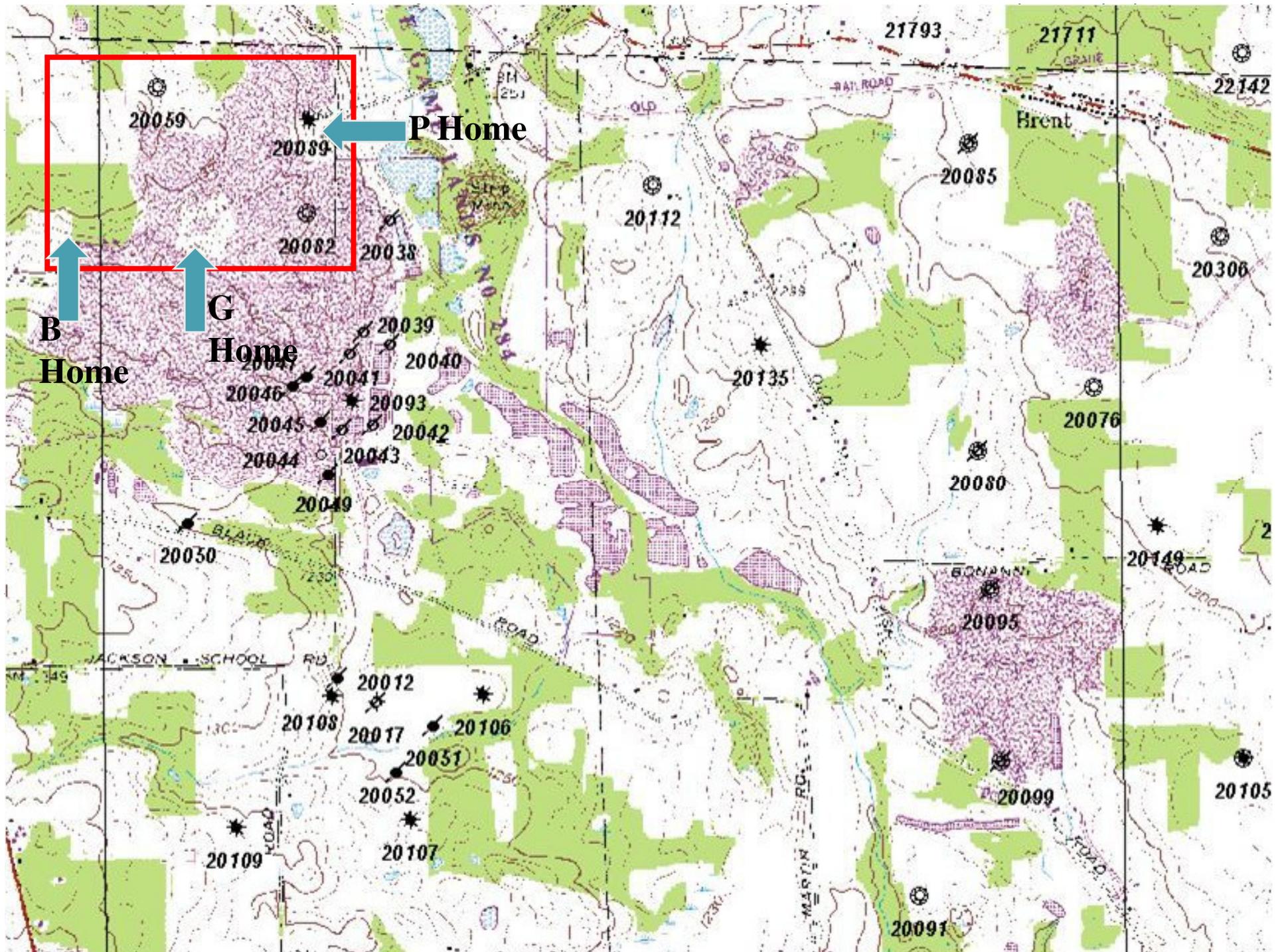
 Limestone

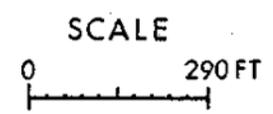
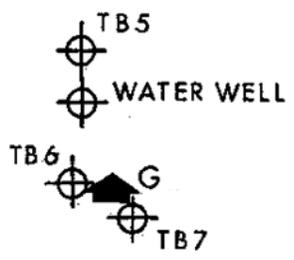
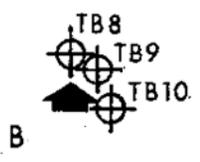
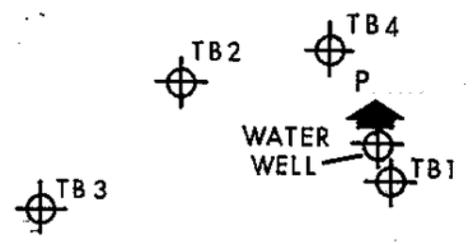
 Coal

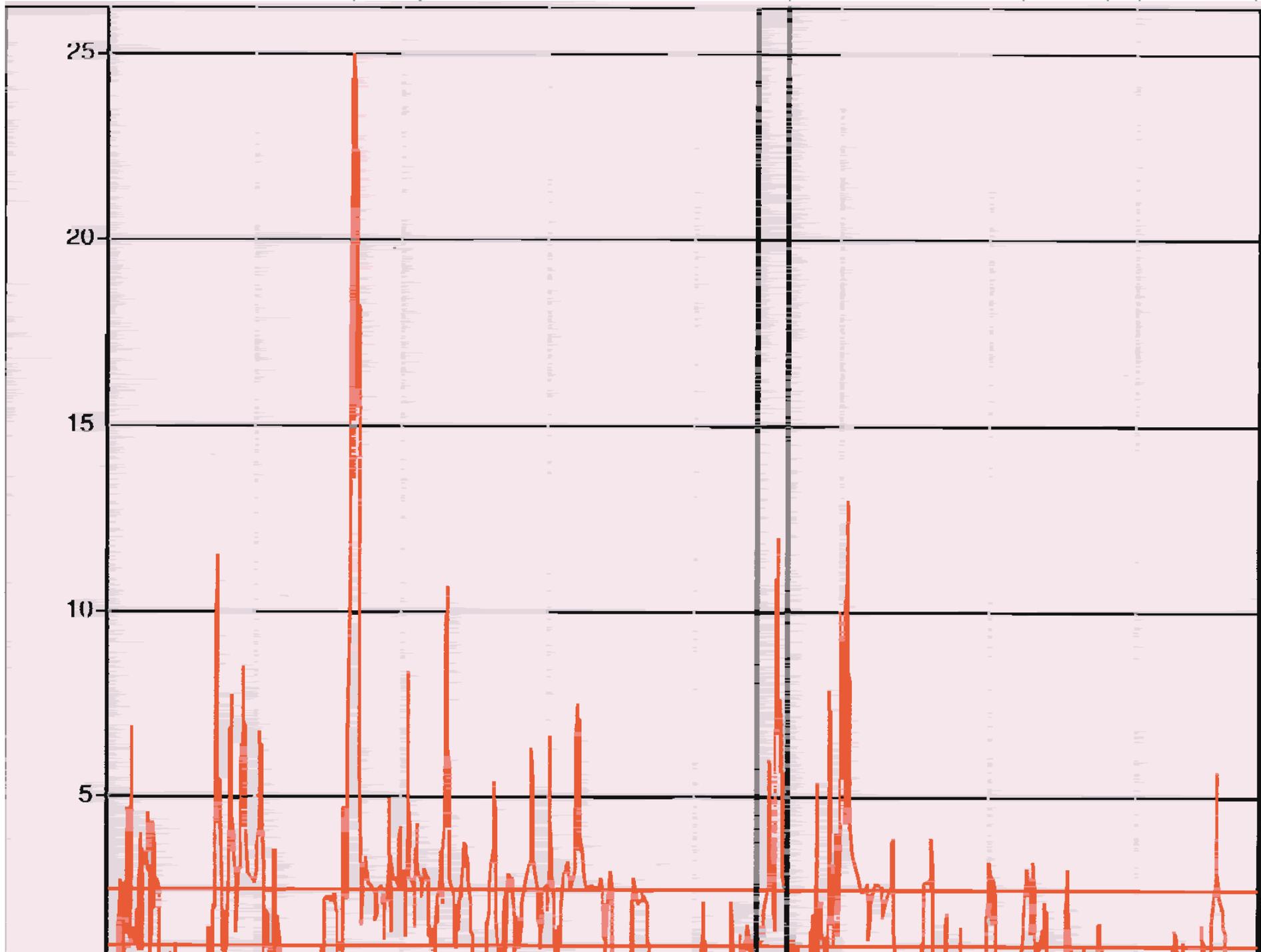
Washington Township, northern Lawrence County

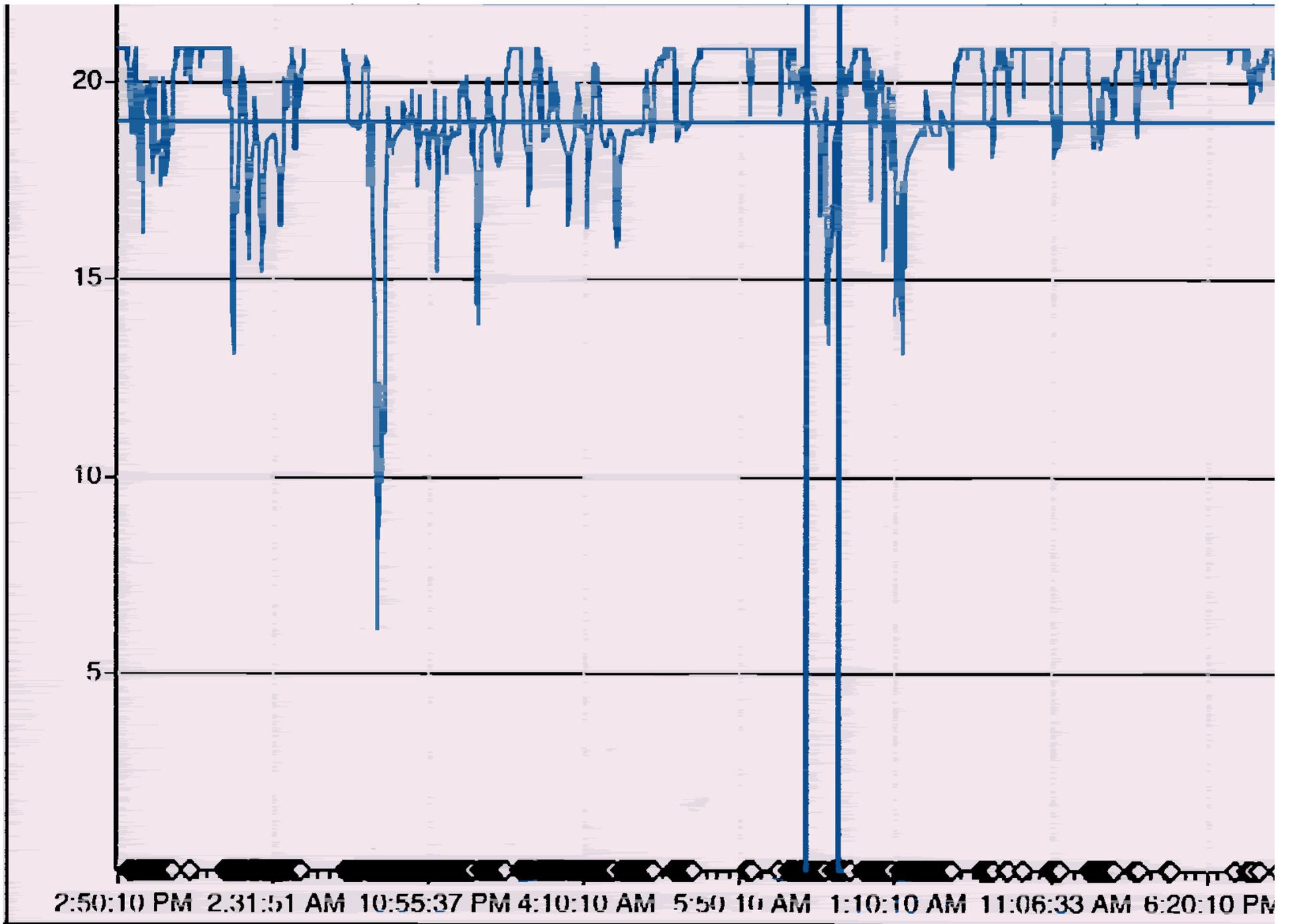
- Three homes
- CO₂ levels > 25% measured inside the homes
- Corresponding O₂ levels depressed as low as 10% inside the homes
- CO₂ levels >25% measured in monitoring wells around the homes
- *Low barometric pressure*
- AML funding





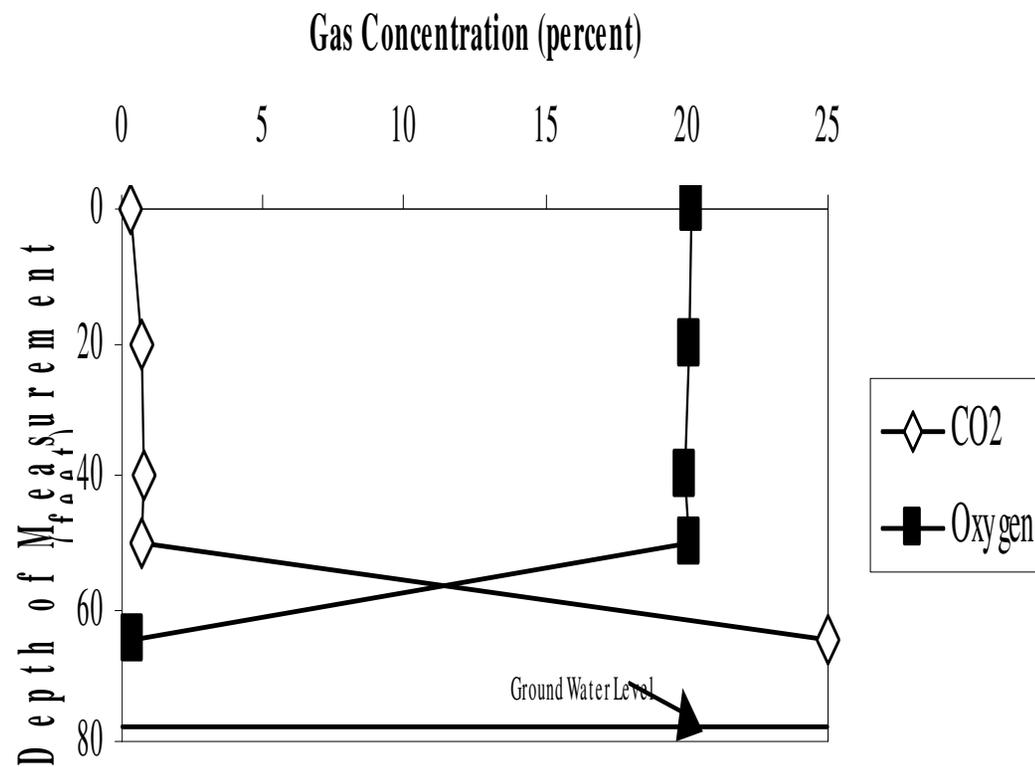






Monitoring Wells

Gas Concentration Profile for DH#2



Lawrence County Potential CO₂ Contamination Sources

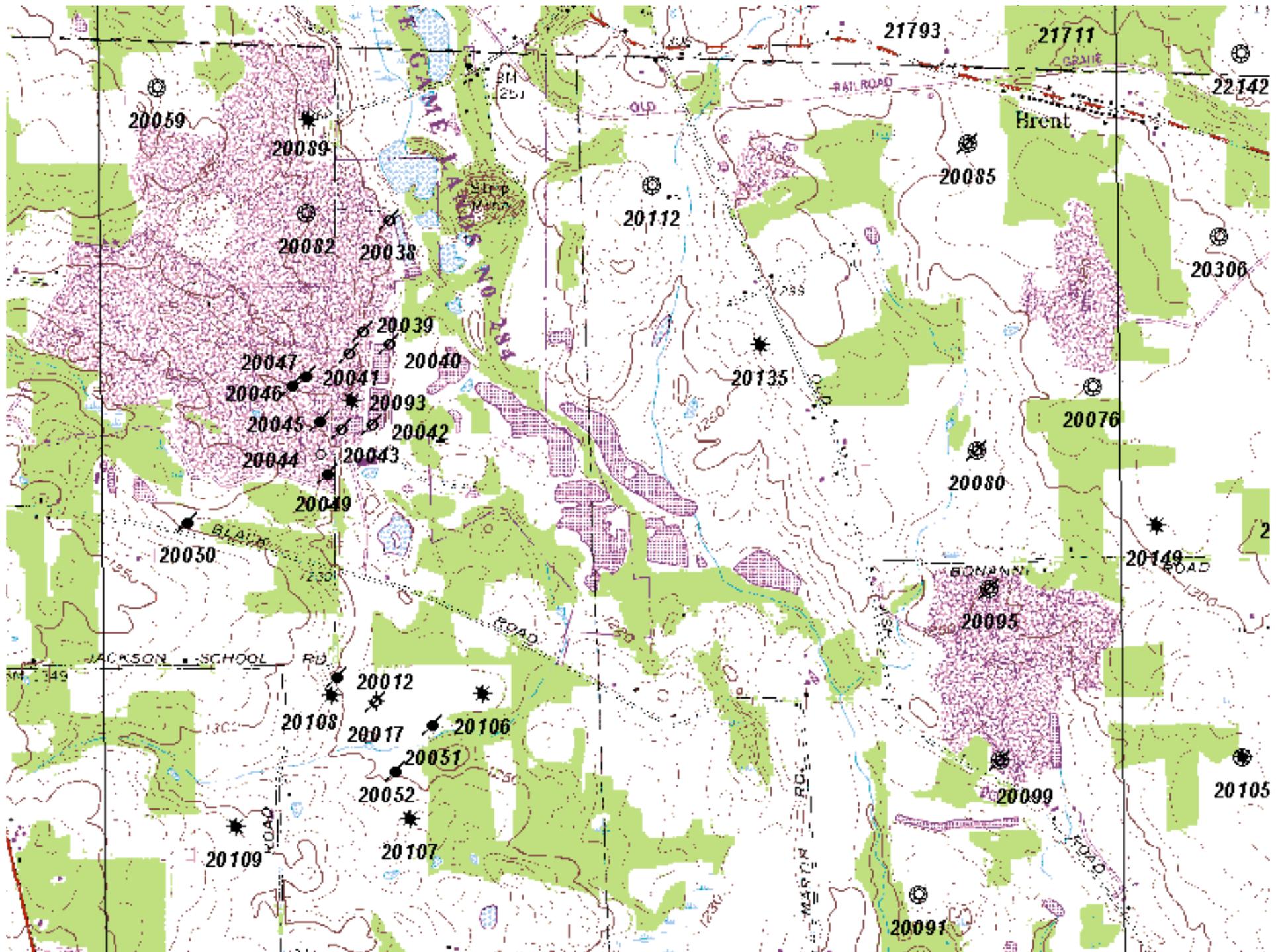
- Deep Mine
- Strip Mines
 - Spoil gas
- Oil and Gas Fields
- Marsh Gas
- Drift Gas
- Subsurface Geology
 - Carbonate rocks?
 - Caves?



Abandoned deep mine

- Brookville coal
- Completely flooded
- Abandoned mine drainage (AMD)
- Discharges up to 2.83 m³/s (100 ft³/s)
- Not a source of black damp



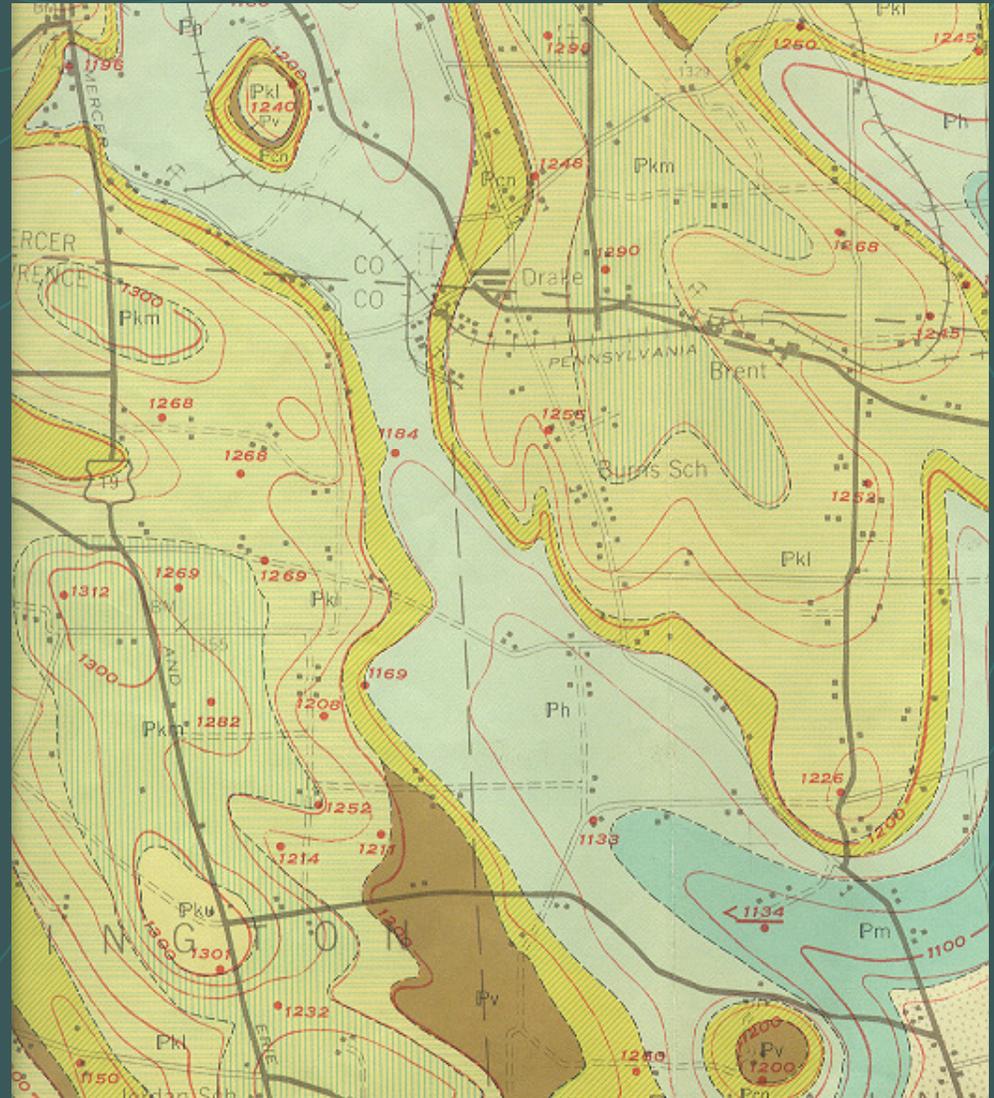






Bedrock Geology

- Allegheny Group
 - Vanport Limestone
 - Cut out by channel ss of the Kittanning Fm in the study location
- Highly fractured bedrock (affects groundwater levels and gas migration)



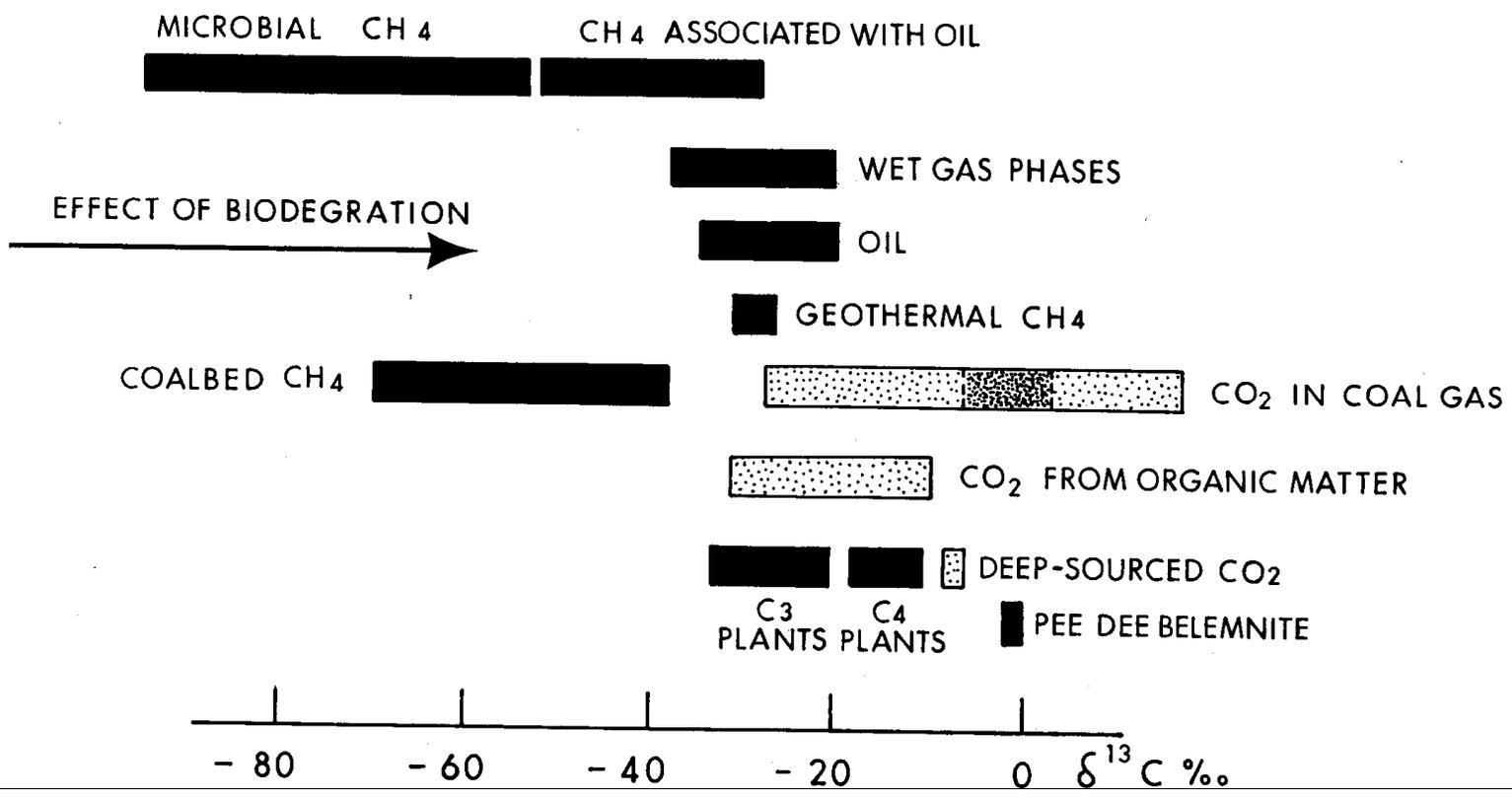
Stable Isotope Geochemistry

$$\delta^{13}\text{CO}_2 = \frac{R_{\text{sample}} - R_{\text{standard}}}{R_{\text{standard}}} * 1000$$

- Isotopic ranges of natural gases are large in range, specific, predictable, and capable of providing diagnostic information on their source.
- Differences in isotopic mass lead to subtle but significant differences in the behavior of an element during natural processes (*fractionation*)

● Carbon

- Highly oxidized compounds (CO₂)
- Highly reduced organic compounds
- Isotope equilibrium exchange reactions – enrichment of ¹³C in carbonate rocks
- Kinetic isotope effects during photosynthesis – concentrate ¹²C in organic matter

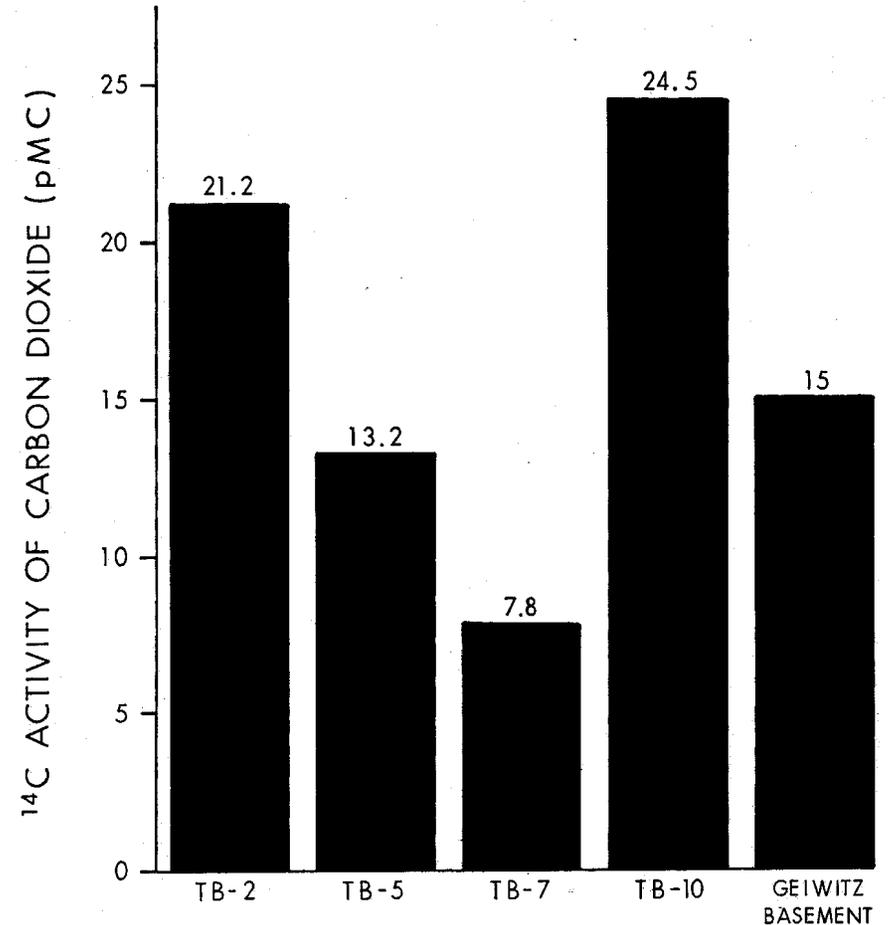
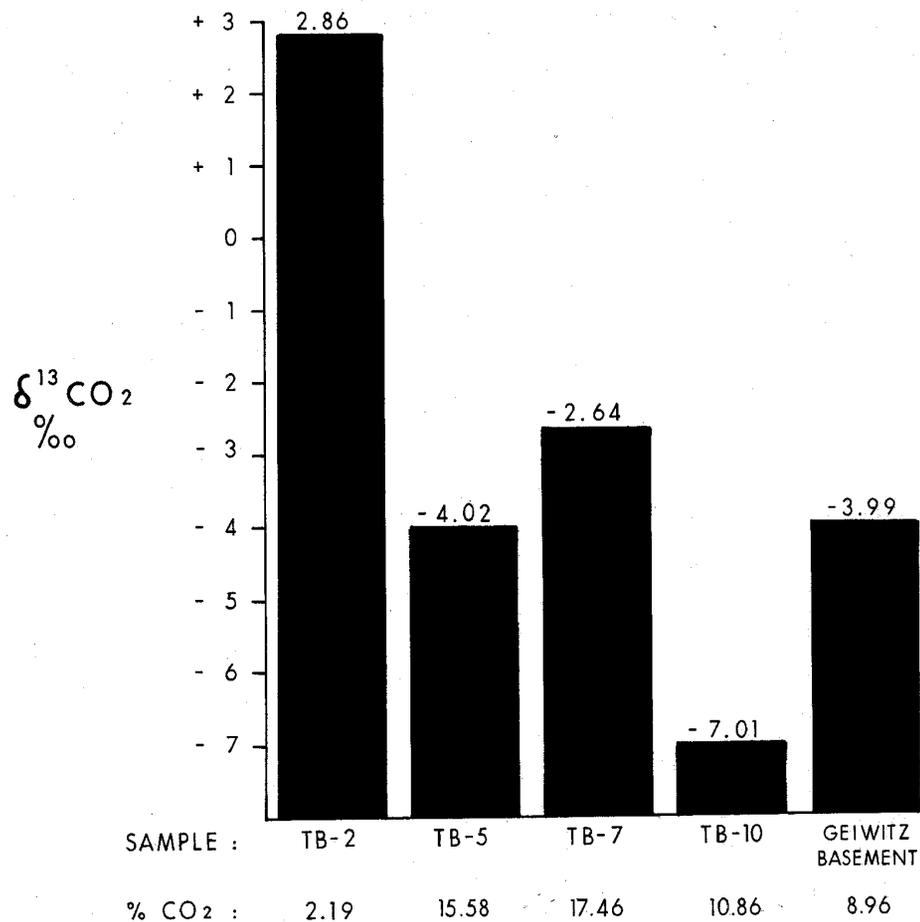


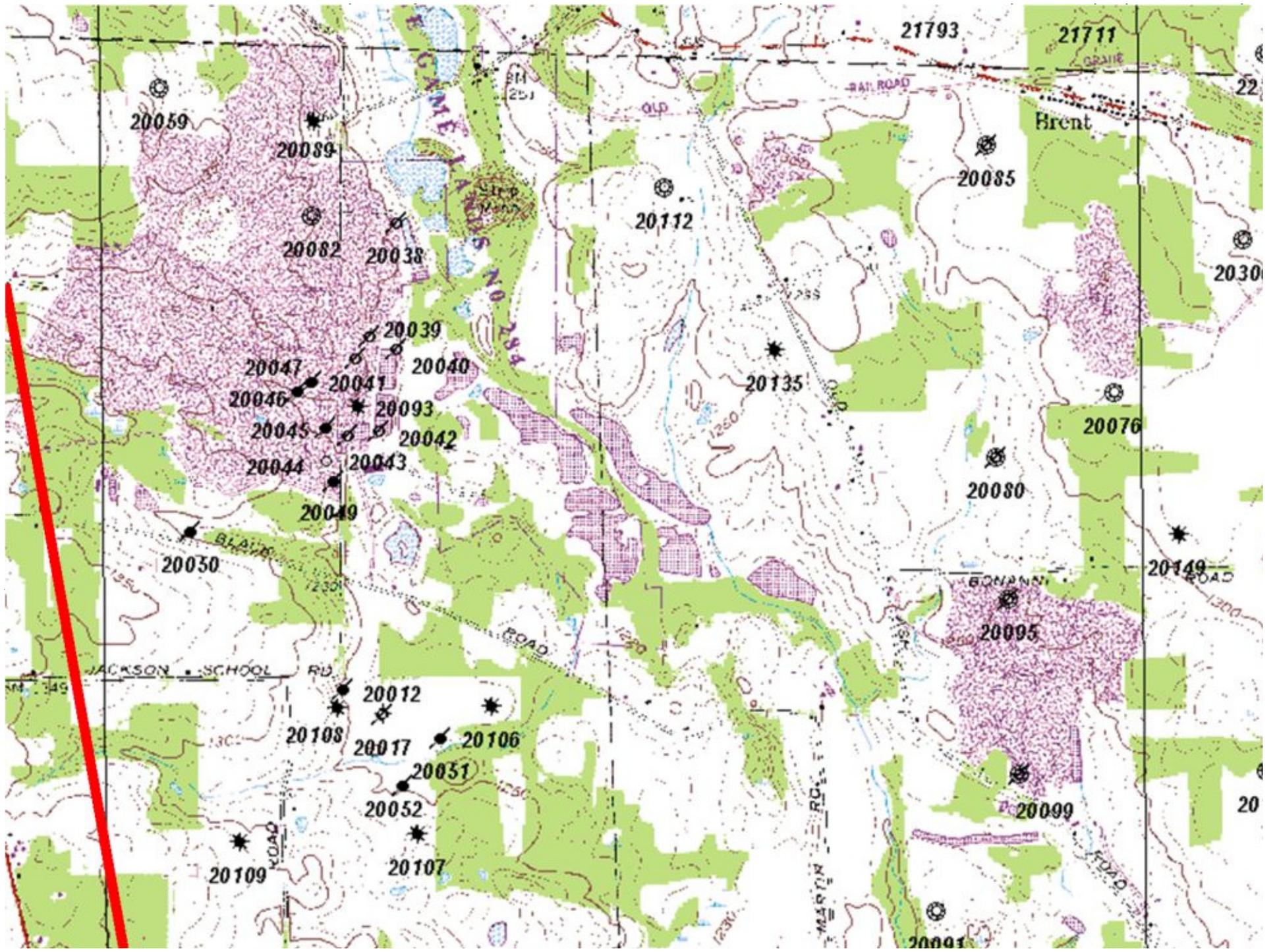
^{14}C Activity of CO_2

- Atmospheric testing of nuclear devices in the 1950's and 1960's produced substantial amounts of ^{14}C -enriched CO_2 .
- All plant or animal material that has grown since approximately 1960 has ^{14}C concentrations above natural levels.
- This ^{14}C concentration can be used as a tracer for microbial gas.
 - ^{14}C activity 30 – 150 percent modern carbon (pMC)
 - 100 pMC defined as natural (equilibrium) pre-bomb ^{14}C concentration of atmospheric CO_2



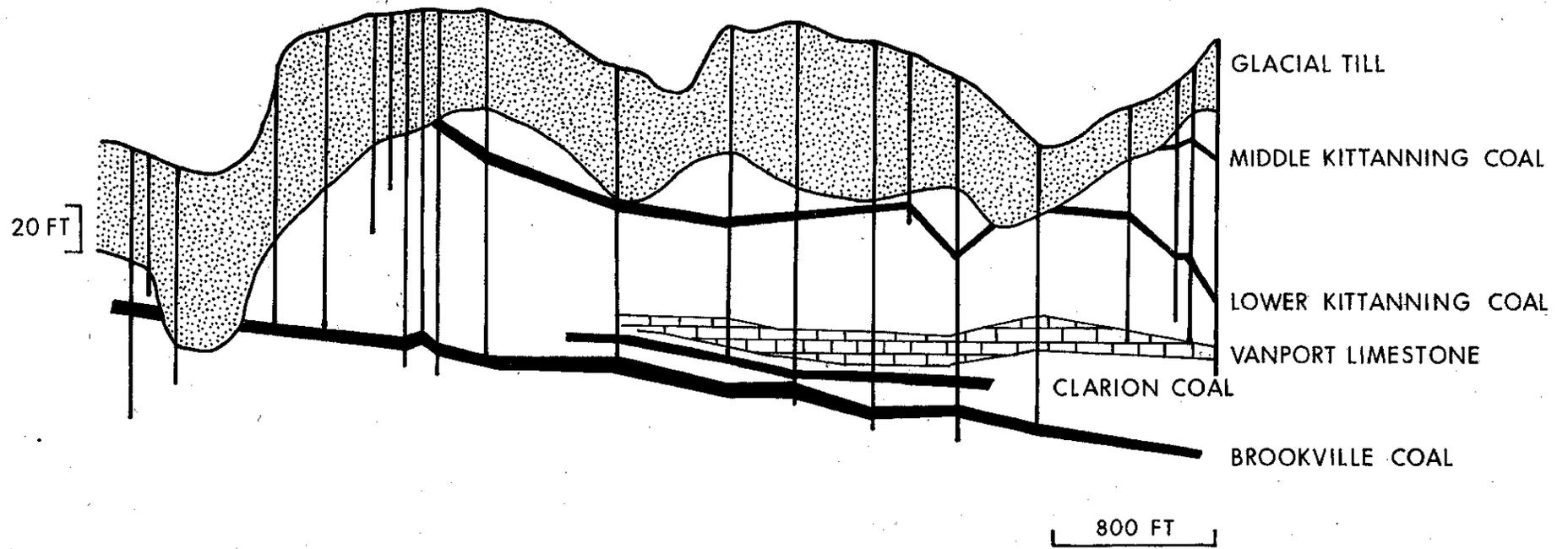
Isotope Geochemistry Data





NORTH

SOUTH



GLACIAL
TILL



COAL



SANDSTONE
AND SHALE



LIMESTONE

Carbonate Source?

- No Vanport Limestone beneath the study area
- Other bedrock units lack carbonate minerals
- Glacial till contains 4 – 8% CaCO_3
- Glacial till was used as fill (high neutralization potential)



Groundwater Chemistry

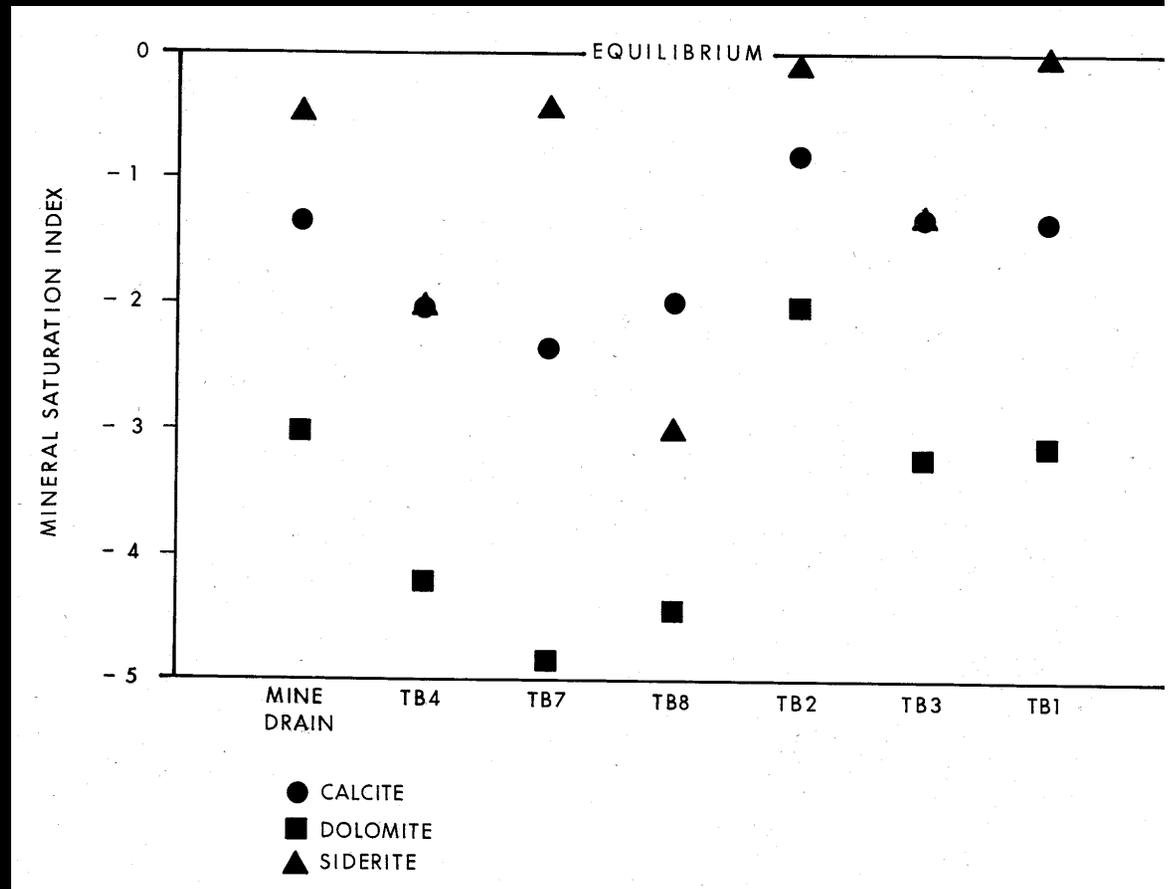
The groundwater onsite is under saturated with respect to calcite and dolomite. Groundwater can react with reported carbonate in the fill to produce the observed volumes of CO_2 .

Alkalinity: 31.36 – 136.84 mg/l

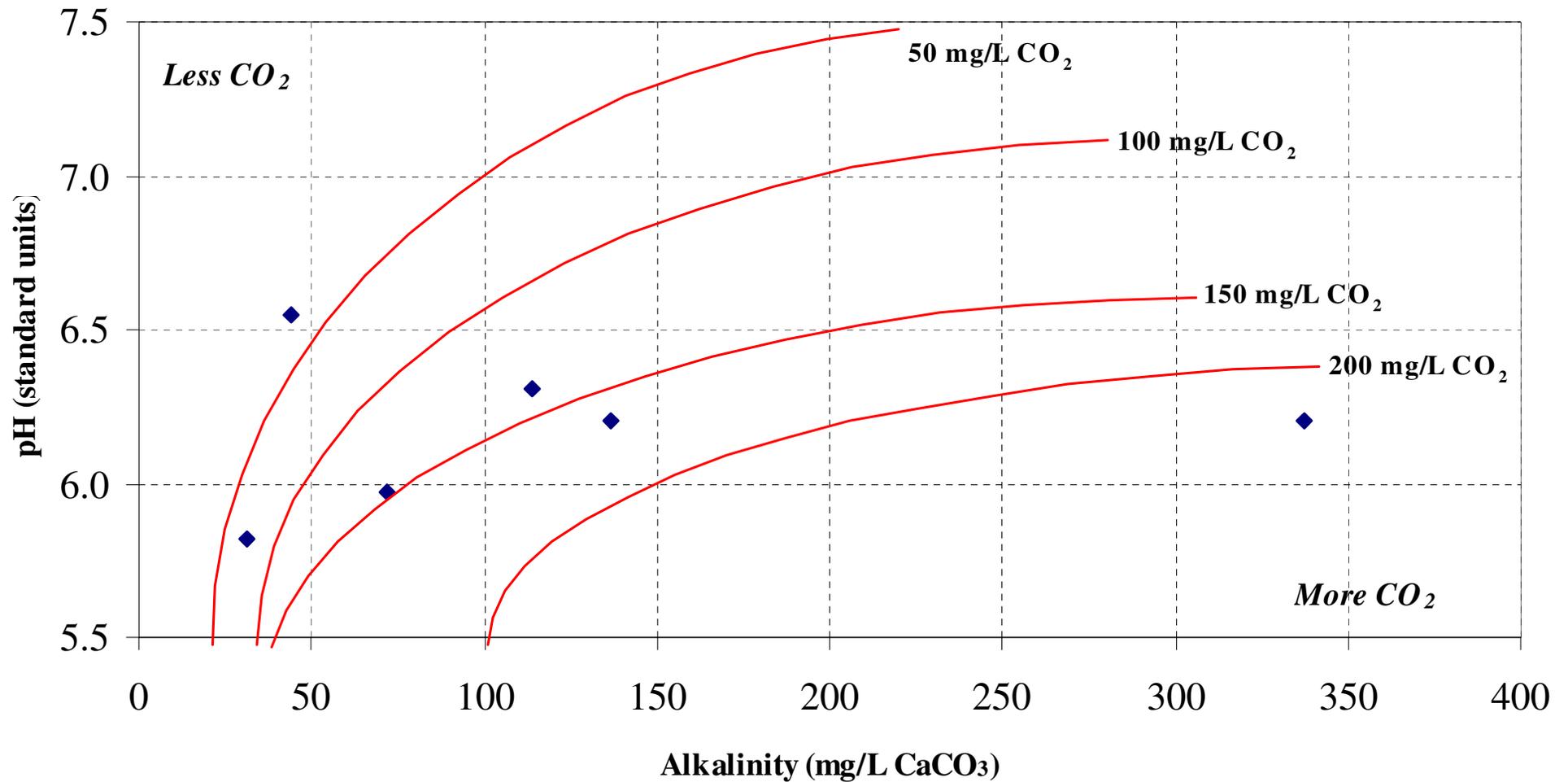
pH: 5.82 – 6.55

PCO_2 : 0.01 – 0.2 atm; groundwater PCO_2 ranged from 0.5 - 0.2 atm during low barometric pressure events.

Local pockets of tippie refuse in fill allow areas of more acidic water and enhance carbonate dissolution.



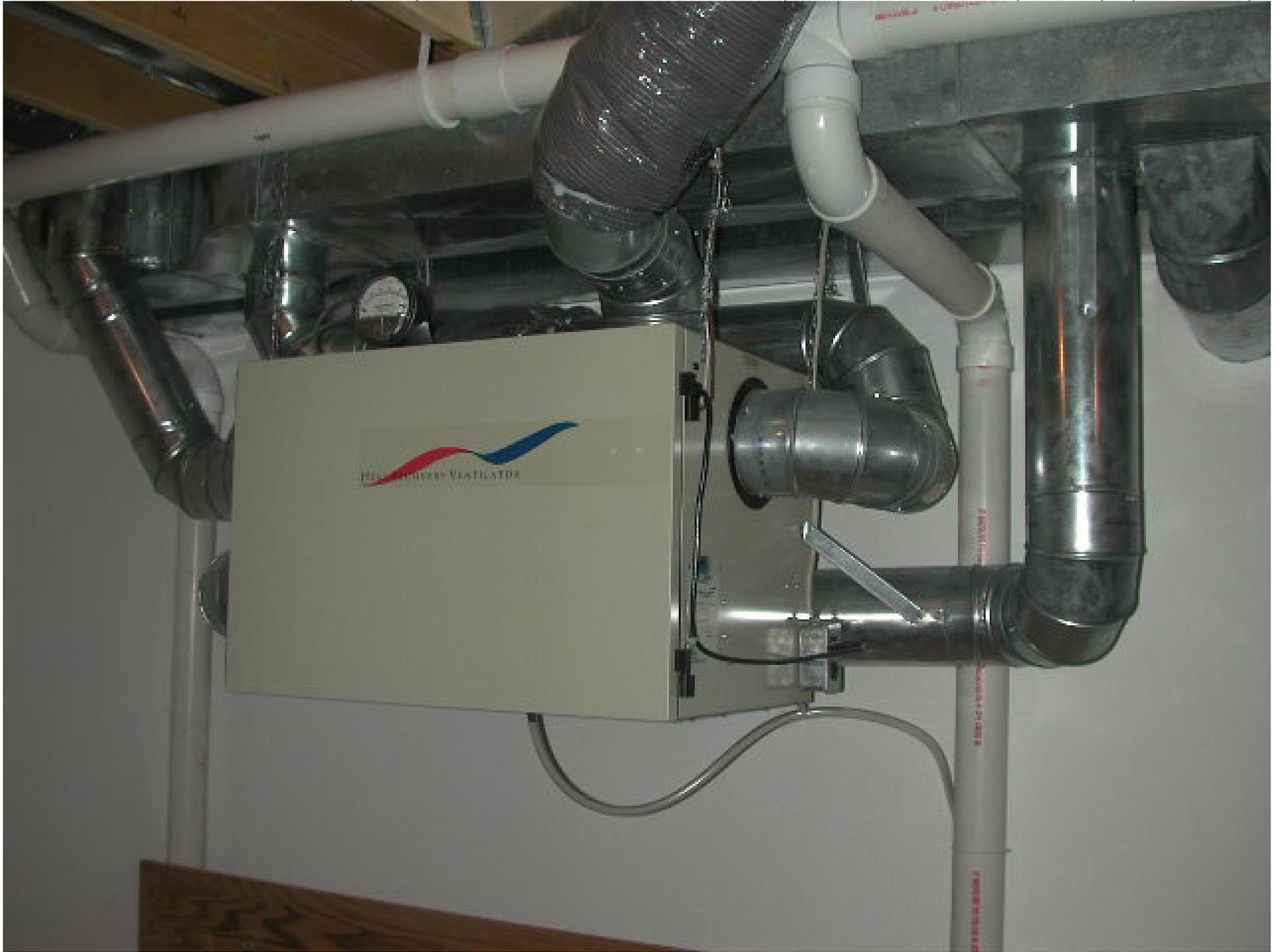
Carbon Dioxide Concentrations at Varying pH and Alkalinity Values

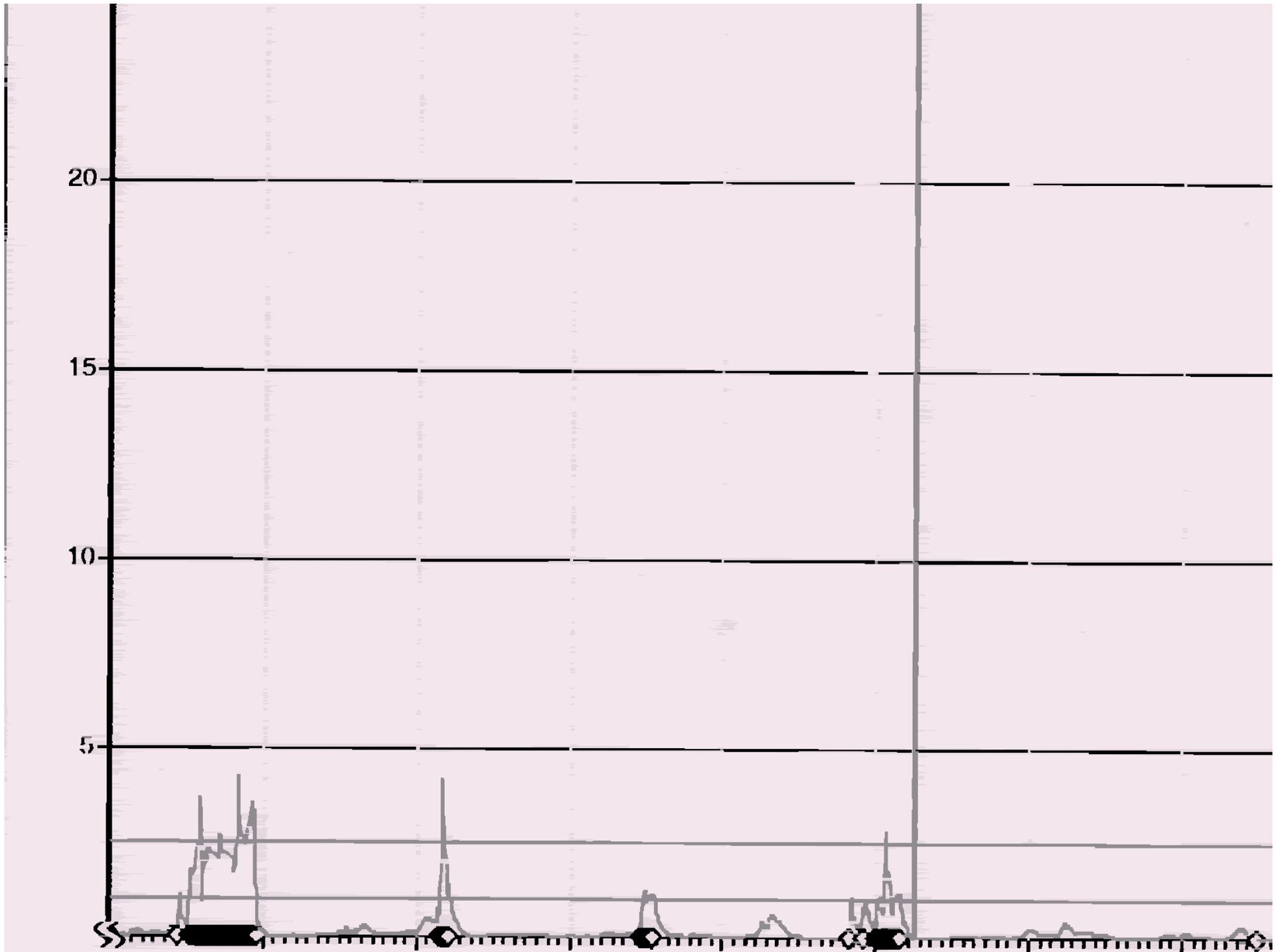












Westmoreland County, Pennsylvania

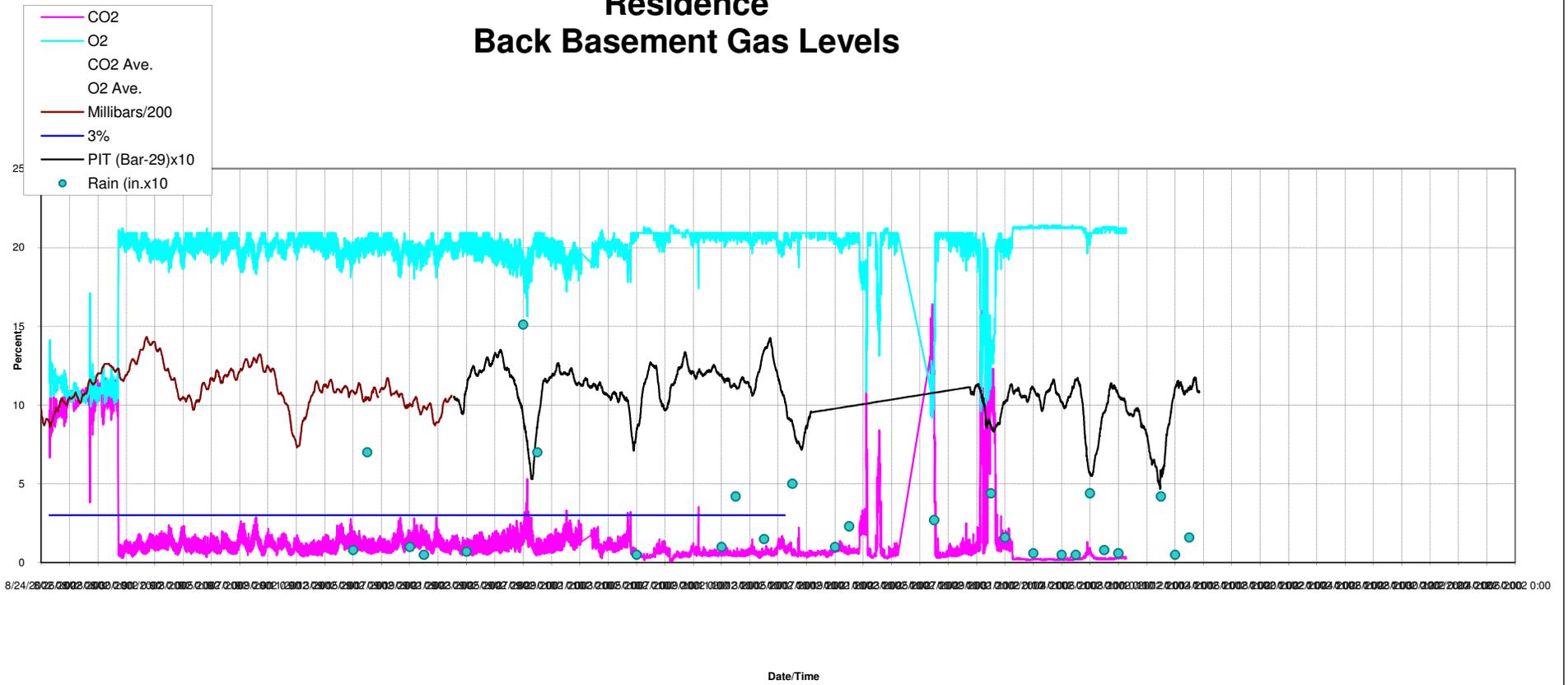
- August 2002: Family members experienced serious symptoms of CO₂ toxicity.
- CO₂ levels as high as 10%
- O₂ levels as low as 15%

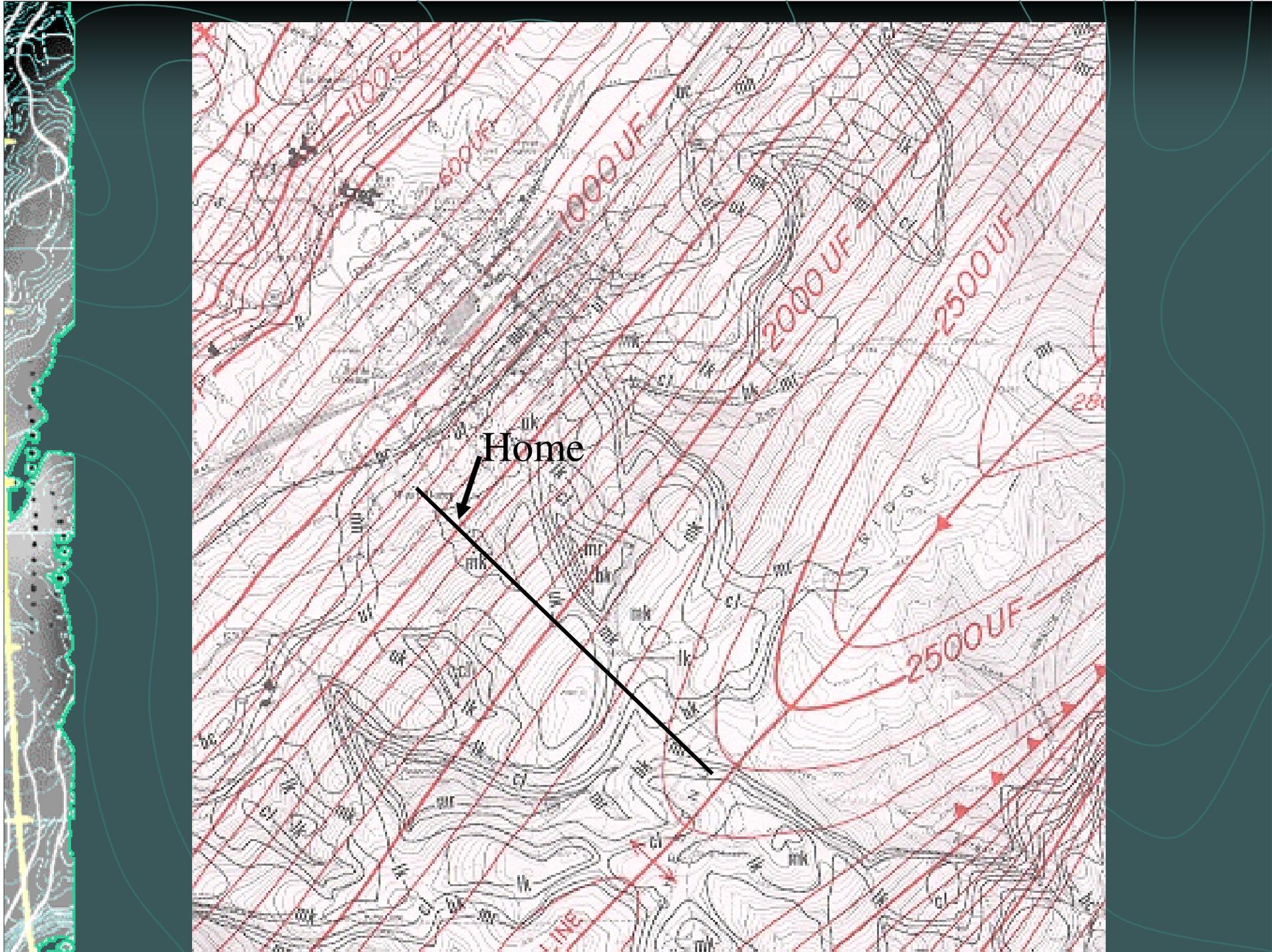


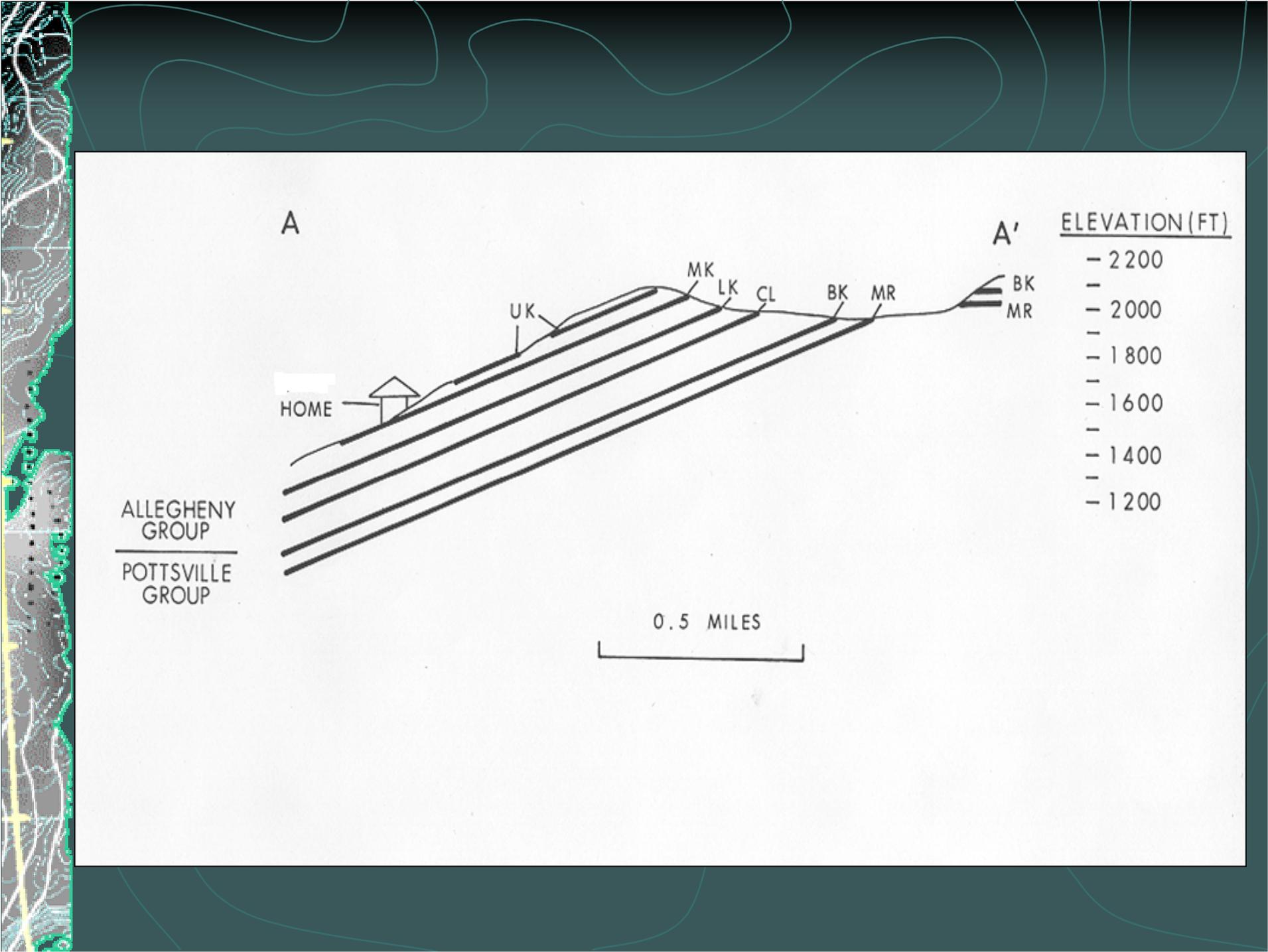


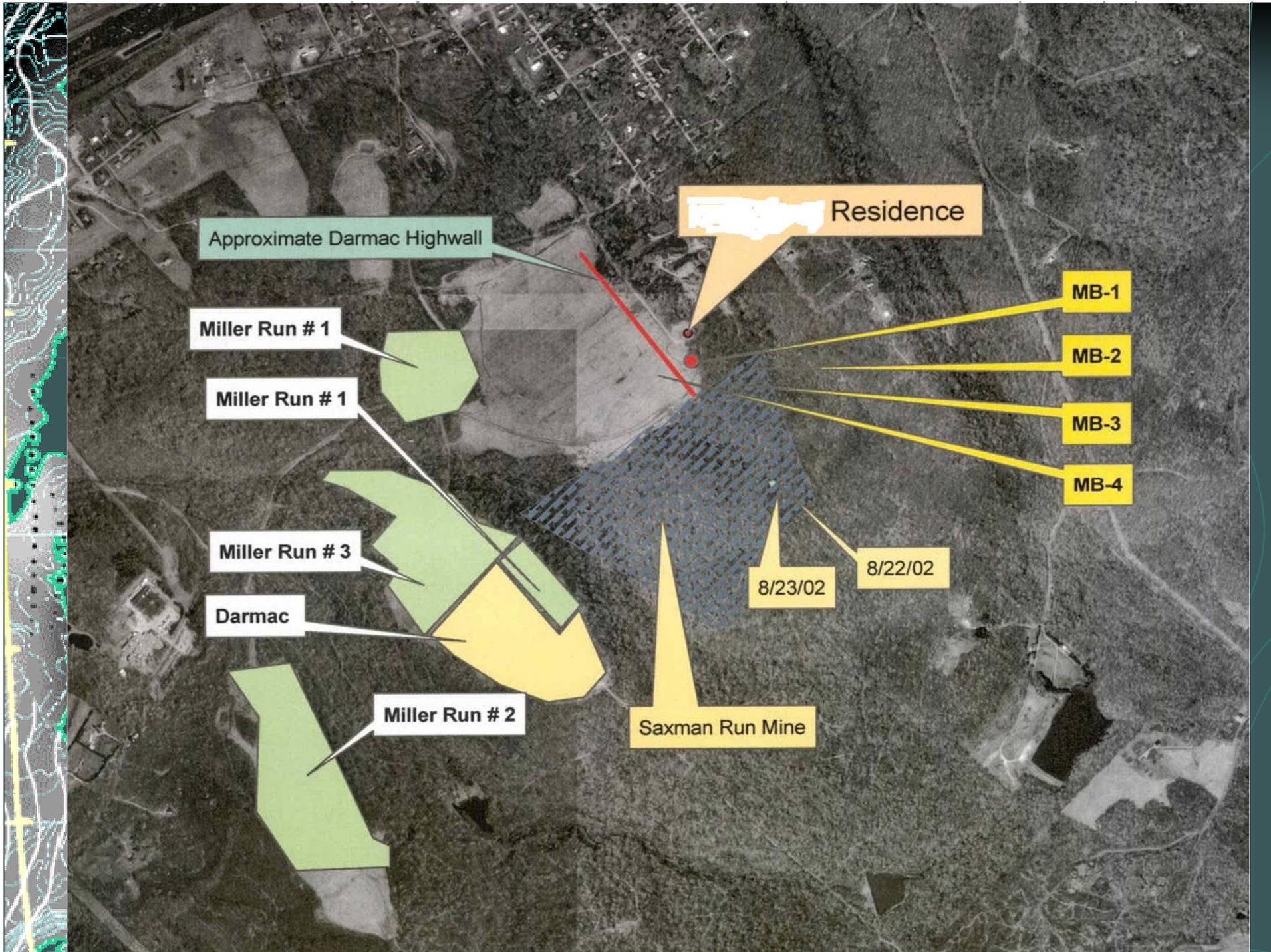


Residence Back Basement Gas Levels









Gas Investigation

Residence

Darmac Highwall

Saxman Run Mine

Active Blast Areas

MB-1

MB-2

MB-3

MB-4

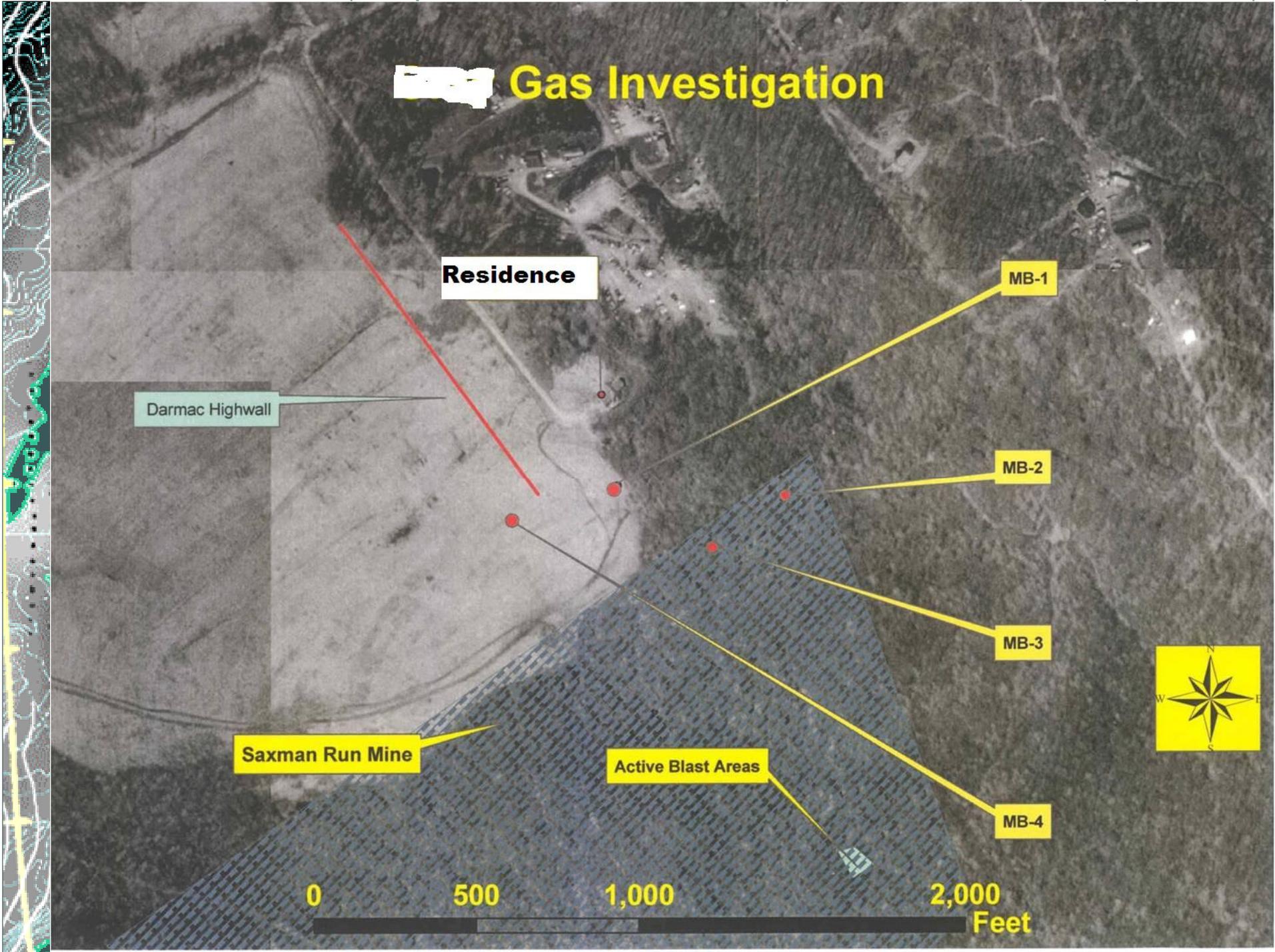
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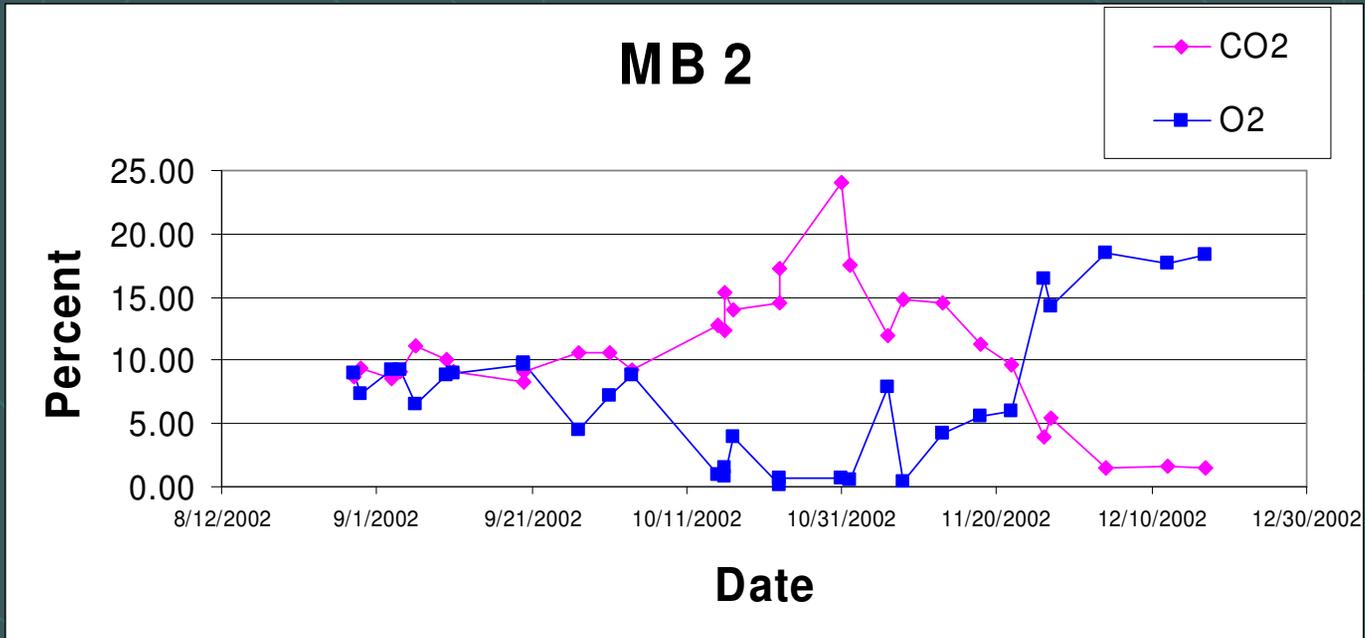
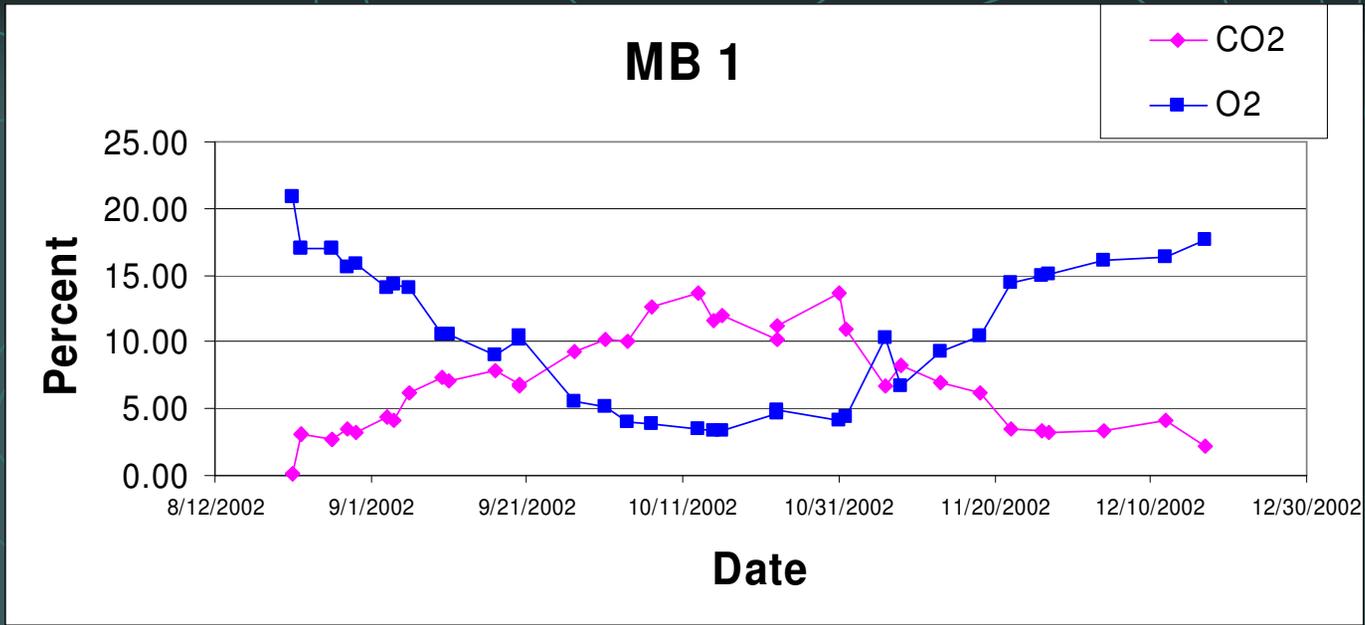
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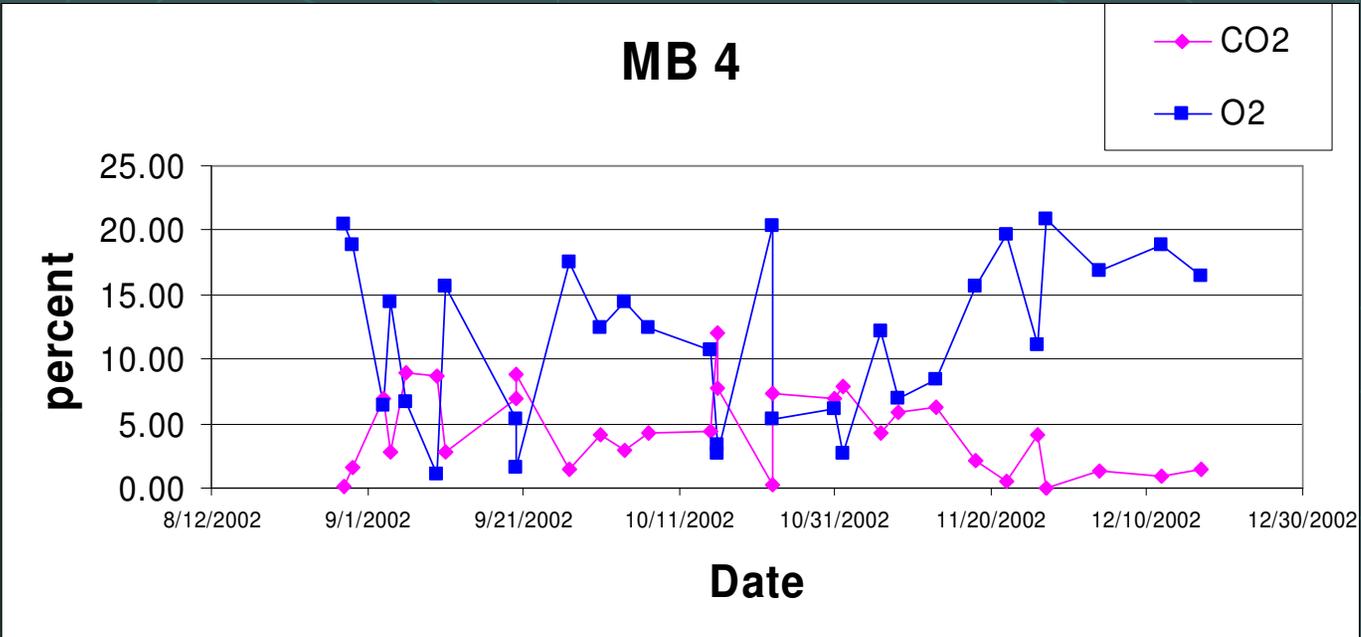
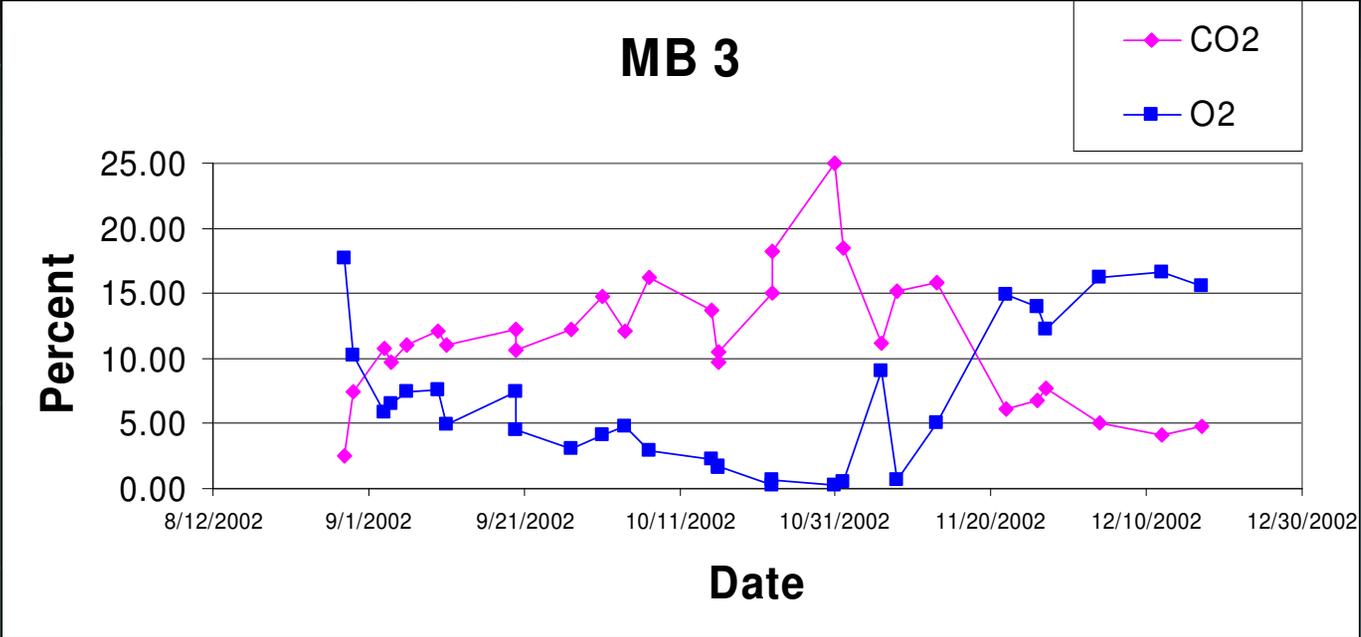
1,000

2,000

Feet



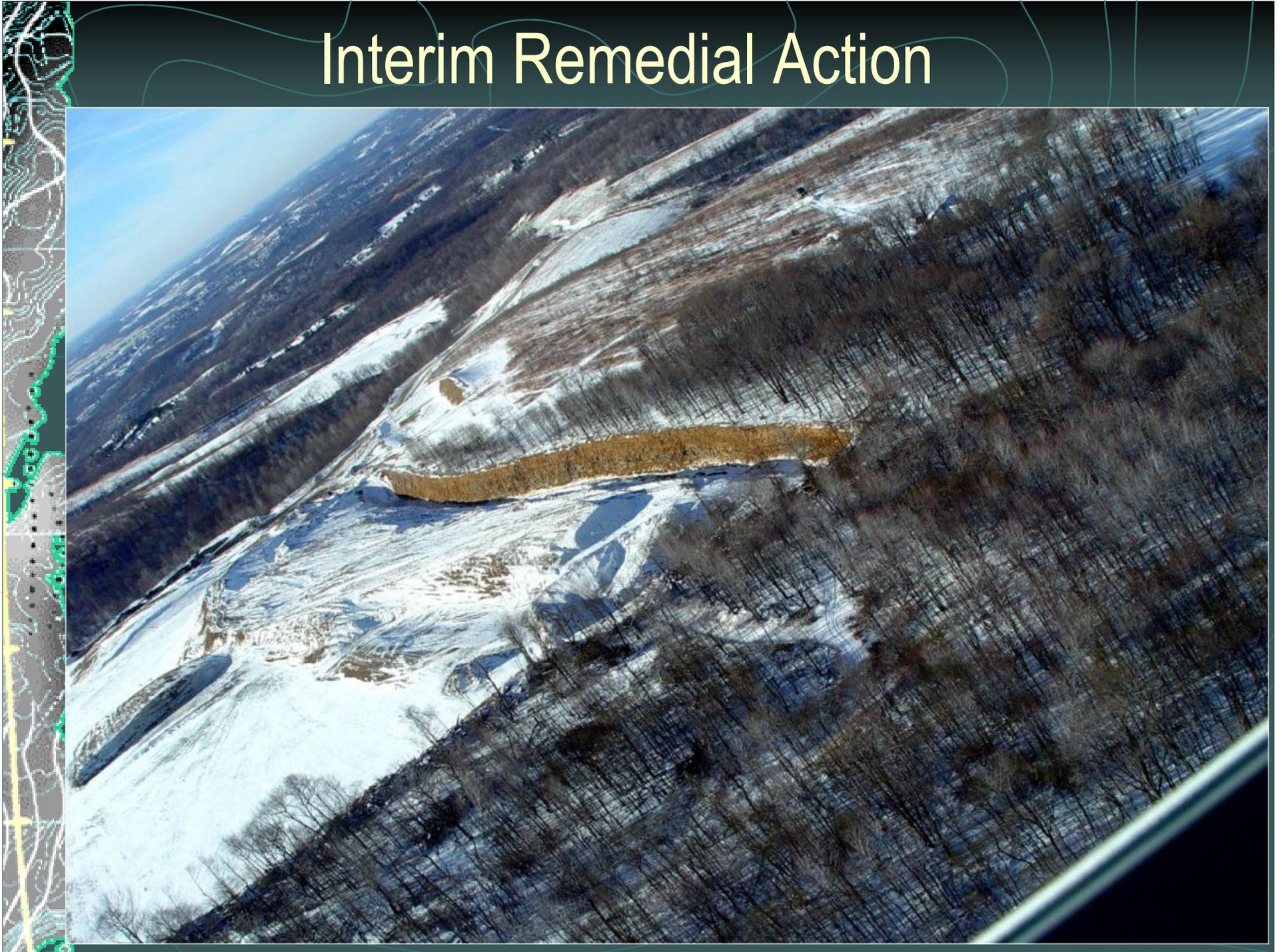




Molecular & Isotopic Results

	Residence	MB-2 (Saxman Run Mine - active)	MB-4 (Darmac Mine - reclaimed)	Crushed Limestone (Used as reclamation at active Saxman Run Mine)
CO ₂	8.86%	10.65%	1.43%	
CH ₄	0.015%	0.013%	0.0087%	
N ₂	80.35%	81.09%	78.00%	
O ₂	9.82%	7.29%	19.30%	
Ar	0.96%	0.96%	0.93%	
N ₂ O	est. at <100 ppm			
C ₃ H ₈			0.17%	
i- C ₄ H ₁₀			0.013%	
n- C ₄ H ₁₀			0.074%	
C ₆ +			0.071%	
δ ¹³ CO	-5.52	-4.76	-23.14	-2.20

Interim Remedial Action





Current stray CO2 cases



Potential for future problems...





CO₂ Migration From Subsurface Sources In Western Pennsylvania

- Relevant to CO₂ sequestration research
- Indicates possible problems related to CO₂ migration
- Exemplifies the myriad sources of CO₂ (“noise”) in potential sequestration sites

